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## Cinnamyl 8-methoxy-2-oxo-2H-chromene-3-carboxylate

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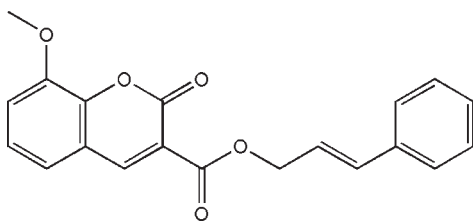
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Key indicators: single-crystal X-ray study;  $T = 296$  K; mean  $\sigma(\text{C}-\text{C}) = 0.004$  Å;  $R$  factor = 0.065;  $wR$  factor = 0.155; data-to-parameter ratio = 12.4.

In the crystal structure of the title compound,  $\text{C}_{20}\text{H}_{16}\text{O}_5$ , the molecule assumes an *E* configuration with the benzene ring and chromenecarboxyl group located on opposite ends of the  $\text{C}=\text{C}$  double bond. The chromene ring system and benzene ring are oriented at a dihedral angle of  $74.66$  ( $12$ )°. Weak intermolecular  $\text{C}-\text{H}\cdots\text{O}$  hydrogen bonding is present in the crystal structure.

## Related literature

For applications of coumarins and related compounds, see: Houlst & Paya (1996); Yu *et al.* (2003, 2007); Finn *et al.* (2004).



## Experimental

## Crystal data

 $\text{C}_{20}\text{H}_{16}\text{O}_5$  $M_r = 336.33$ 

Monoclinic,  $P2_1/c$   
 $a = 19.226$  (4) Å  
 $b = 9.5483$  (19) Å  
 $c = 9.0046$  (18) Å  
 $\beta = 90.97$  (3)°  
 $V = 1652.8$  (6) Å<sup>3</sup>

$Z = 4$   
 Mo  $K\alpha$  radiation  
 $\mu = 0.10$  mm<sup>-1</sup>  
 $T = 296$  K  
 $0.20 \times 0.17 \times 0.17$  mm

## Data collection

Bruker SMART CCD area detector  
 diffractometer  
 Absorption correction: none  
 4903 measured reflections

2834 independent reflections  
 2157 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.053$

## Refinement

$R[F^2 > 2\sigma(F^2)] = 0.065$   
 $wR(F^2) = 0.155$   
 $S = 1.15$   
 2834 reflections

228 parameters  
 H-atom parameters constrained  
 $\Delta\rho_{\text{max}} = 0.24$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.20$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

| $D-H\cdots A$                                | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|--|-------|-------------|-------------|---------------|
| $\text{C4}-\text{H4A}\cdots\text{O3}^i$      | 0.93  | 2.51        | 3.429 (3)   | 170           |
| $\text{C17}-\text{H17A}\cdots\text{O4}^{ii}$ | 0.93  | 2.44        | 3.294 (4)   | 153           |

Symmetry codes: (i)  $x, y + 1, z$ ; (ii)  $-x + 1, y + \frac{1}{2}, -z + \frac{1}{2}$ .

Data collection: *SMART* (Bruker, 2007); cell refinement: *SAINT* (Bruker, 2007); data reduction: *SAINT*; program(s) used to solve structure: *SHELXTL* (Sheldrick, 2008); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: XU2613).

## References

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**supplementary materials**

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## Cinnamyl 8-methoxy-2-oxo-2H-chromene-3-carboxylate

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### Comment

Coumarins and related compounds, kinds of plant-derived compounds, have diverse biological activities, including anti-HIV, anti-bacterial, anti-inflammatory, anti-proliferative and antioxidant properties (Hoult & Paya, 1996; Yu *et al.*, 2003; Finn *et al.*, 2004; Yu *et al.*, 2007). It thus appeared of interest to synthesize the compounds with coumarin-skeleton. As part of work, we have synthesized the title compound (I) and report its crystal structure here.

The title molecule crystallizes in the E conformation, with an C12-C13-C14-C15 torsion angle of  $-179.5(3)^\circ$ . The 8-methoxy-2H-chromen-2-one ring and the C15-benzene ring make a dihedral of  $74.66(12)^\circ$ .

In the crystal structure, an intramolecular C—H $\cdots$ O hydrogen bond is observed and helps to stabilize the conformation of the molecule.

### Experimental

A solution of cinnamic alcohol (7.2 mmol) dissolved in dried methyl dichloride (DCM) (25 ml) was added dropwise to a solution of 2-oxo-2H-chromene-3-acyl chloride (7.2 mmol) dissolved in DCM (25 ml) and triethylamine (1 ml) at room temperature. The reaction mixture was stirred for 24 h, monitored by TLC. The reaction mixture was neutralized with 5% HCl and washed by saturated NaHCO<sub>3</sub> solution and brine, respectively. The organic phase is dried over Na<sub>2</sub>SO<sub>4</sub> and evaporated under the reduced pressure. The resulting residue was purified by column chromatography (EtOAc:petroleum ether) to give the purified compound.

### Refinement

All H atoms were positioned geometrically and refined as riding with C—H = 0.93 (aromatic), 0.97 (methylene) and 0.96 Å (methyl),  $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$  for methyl H atoms and  $1.2U_{\text{eq}}(\text{C})$  for the others.

### Figures

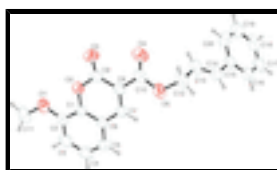


Fig. 1. The molecular structure of the title compound. Displacement ellipsoids are drawn at the 50% probability level for non-H atoms.

## cinnamyl 8-methoxy-2-oxo-2H-chromene-3-carboxylate

### Crystal data

|                                |   |
|--------------------------------|---|
| $C_{20}H_{16}O_5$              | $F_{000} = 704$   |
| $M_r = 336.33$                 | $D_x = 1.352 \text{ Mg m}^{-3}$                         |
| Monoclinic, $P2_1/c$           | Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$ |
| Hall symbol: -P 2ybc           | Cell parameters from 2834 reflections                   |
| $a = 19.226 (4) \text{ \AA}$   | $\theta = 3.1\text{--}24.2^\circ$                       |
| $b = 9.5483 (19) \text{ \AA}$  | $\mu = 0.10 \text{ mm}^{-1}$                            |
| $c = 9.0046 (18) \text{ \AA}$  | $T = 296 \text{ K}$                                     |
| $\beta = 90.97 (3)^\circ$      | Block, colorless  |
| $V = 1652.8 (6) \text{ \AA}^3$ | $0.20 \times 0.17 \times 0.17 \text{ mm}$               |
| $Z = 4$                        |   |

### Data collection

|  |  |
|--|--|
| Bruker SMART CCD area detector<br>diffractometer | 2157 reflections with $I > 2\sigma(I)$ |
| Radiation source: fine-focus sealed tube         | $R_{\text{int}} = 0.053$               |
| Monochromator: graphite                          | $\theta_{\text{max}} = 25.0^\circ$     |
| $T = 296 \text{ K}$                              | $\theta_{\text{min}} = 1.1^\circ$      |
| $\omega$ scans                                   | $h = -22 \rightarrow 22$               |
| Absorption correction: none                      | $k = -11 \rightarrow 11$               |
| 4903 measured reflections                        | $l = 0 \rightarrow 10$                 |
| 2834 independent reflections                     |  |

### Refinement

|  |   |
|--|---|
| Refinement on $F^2$  | Hydrogen site location: inferred from neighbouring sites                                    |
| Least-squares matrix: full                                     | H-atom parameters constrained   |
| $R[F^2 > 2\sigma(F^2)] = 0.065$                                | $w = 1/[\sigma^2(F_o^2) + (0.0659P)^2 + 0.2012P]$   |
| $wR(F^2) = 0.155$  | where $P = (F_o^2 + 2F_c^2)/3$  |
| $S = 1.15$   | $(\Delta/\sigma)_{\text{max}} = 0.001$  |
| 2834 reflections   | $\Delta\rho_{\text{max}} = 0.24 \text{ e \AA}^{-3}$   |
| 228 parameters   | $\Delta\rho_{\text{min}} = -0.20 \text{ e \AA}^{-3}$  |
| Primary atom site location: structure-invariant direct methods | Extinction correction: SHELXL,<br>$F_c^* = kFc[1+0.001xFc^2\lambda^3/\sin(2\theta)]^{-1/4}$ |
| Secondary atom site location: difference Fourier map           | Extinction coefficient: 0.024 (3)   |

*Special details*

**Geometry.** All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -factors(gt) etc. and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

|      | <i>x</i>      | <i>y</i>     | <i>z</i>     | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|------|---------------|--------------|--------------|----------------------------------|
| O1   | 0.04001 (9)   | 0.88854 (18) | 0.70962 (19) | 0.0610 (5)                       |
| O2   | 0.11713 (7)   | 0.72996 (15) | 0.54027 (17) | 0.0463 (4)                       |
| O3   | 0.15130 (10)  | 0.51497 (17) | 0.4995 (2)   | 0.0705 (6)                       |
| O4   | 0.25790 (11)  | 0.4732 (2)   | 0.3000 (3)   | 0.0885 (7)                       |
| O5   | 0.28952 (8)   | 0.66346 (19) | 0.17300 (19) | 0.0617 (5)                       |
| C1   | 0.12199 (11)  | 0.8722 (2)   | 0.5210 (2)   | 0.0414 (6)                       |
| C2   | 0.08113 (11)  | 0.9568 (3)   | 0.6115 (3)   | 0.0497 (6)                       |
| C3   | 0.08597 (14)  | 1.0996 (3)   | 0.5946 (3)   | 0.0652 (8)                       |
| H3A  | 0.0593        | 1.1578       | 0.6536       | 0.078*                           |
| C4   | 0.13001 (15)  | 1.1585 (3)   | 0.4910 (3)   | 0.0710 (8)                       |
| H4A  | 0.1323        | 1.2554       | 0.4816       | 0.085*                           |
| C5   | 0.17013 (13)  | 1.0757 (3)   | 0.4023 (3)   | 0.0605 (7)                       |
| H5A  | 0.1992        | 1.1160       | 0.3328       | 0.073*                           |
| C6   | 0.16698 (11)  | 0.9293 (2)   | 0.4174 (3)   | 0.0445 (6)                       |
| C7   | 0.20856 (11)  | 0.8339 (2)   | 0.3351 (2)   | 0.0450 (6)                       |
| H7A  | 0.2383        | 0.8693       | 0.2638       | 0.054*                           |
| C8   | 0.20641 (10)  | 0.6940 (2)   | 0.3568 (2)   | 0.0429 (6)                       |
| C9   | 0.15868 (11)  | 0.6355 (2)   | 0.4665 (3)   | 0.0460 (6)                       |
| C10  | 0.25260 (12)  | 0.5959 (3)   | 0.2770 (3)   | 0.0525 (6)                       |
| C11  | -0.00432 (15) | 0.9739 (3)   | 0.7994 (3)   | 0.0743 (8)                       |
| H11A | -0.0314       | 0.9150       | 0.8626       | 0.112*                           |
| H11B | -0.0349       | 1.0278       | 0.7362       | 0.112*                           |
| H11C | 0.0237        | 1.0360       | 0.8592       | 0.112*                           |
| C12  | 0.33877 (13)  | 0.5785 (3)   | 0.0894 (3)   | 0.0697 (8)                       |
| H12A | 0.3470        | 0.6224       | -0.0059      | 0.084*                           |
| H12B | 0.3189        | 0.4867       | 0.0709       | 0.084*                           |
| C13  | 0.40569 (13)  | 0.5629 (3)   | 0.1714 (3)   | 0.0646 (7)                       |
| H13A | 0.4053        | 0.5142       | 0.2608       | 0.078*                           |
| C14  | 0.46490 (14)  | 0.6123 (3)   | 0.1279 (3)   | 0.0639 (7)                       |
| H14A | 0.4640        | 0.6613       | 0.0386       | 0.077*                           |
| C15  | 0.53343 (12)  | 0.6000 (3)   | 0.2031 (3)   | 0.0560 (7)                       |
| C16  | 0.58725 (14)  | 0.6863 (3)   | 0.1617 (4)   | 0.0741 (9)                       |
| H16A | 0.5804        | 0.7481       | 0.0830       | 0.089*                           |

## supplementary materials

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|      |              |            |            |             |
|------|--------------|------------|------------|-------------|
| C17  | 0.65113 (15) | 0.6830 (3) | 0.2345 (5) | 0.0831 (10) |
| H17A | 0.6861       | 0.7445     | 0.2067     | 0.100*      |
| C18  | 0.66314 (14) | 0.5891 (3) | 0.3481 (4) | 0.0741 (9)  |
| H18A | 0.7059       | 0.5874     | 0.3980     | 0.089*      |
| C19  | 0.61138 (15) | 0.4981 (4) | 0.3866 (3) | 0.0763 (9)  |
| H19A | 0.6195       | 0.4320     | 0.4608     | 0.092*      |
| C20  | 0.54716 (14) | 0.5041 (3) | 0.3158 (3) | 0.0694 (8)  |
| H20A | 0.5123       | 0.4426     | 0.3443     | 0.083*      |

### Atomic displacement parameters ( $\text{\AA}^2$ )

|     | $U^{11}$    | $U^{22}$    | $U^{33}$    | $U^{12}$     | $U^{13}$     | $U^{23}$     |
|-----|-------------|-------------|-------------|--------------|--------------|--------------|
| O1  | 0.0600 (10) | 0.0639 (11) | 0.0597 (11) | 0.0079 (9)   | 0.0171 (8)   | -0.0059 (9)  |
| O2  | 0.0487 (9)  | 0.0410 (9)  | 0.0494 (10) | 0.0015 (7)   | 0.0053 (7)   | -0.0008 (7)  |
| O3  | 0.0869 (14) | 0.0375 (10) | 0.0877 (14) | 0.0016 (9)   | 0.0227 (10)  | 0.0041 (9)   |
| O4  | 0.1028 (16) | 0.0562 (13) | 0.1075 (17) | 0.0296 (11)  | 0.0322 (13)  | 0.0052 (11)  |
| O5  | 0.0522 (10) | 0.0719 (12) | 0.0615 (12) | 0.0113 (9)   | 0.0125 (8)   | -0.0069 (9)  |
| C1  | 0.0391 (12) | 0.0364 (12) | 0.0486 (14) | 0.0004 (10)  | -0.0026 (10) | -0.0021 (10) |
| C2  | 0.0444 (13) | 0.0505 (15) | 0.0541 (15) | 0.0053 (11)  | -0.0019 (11) | -0.0051 (12) |
| C3  | 0.0668 (17) | 0.0523 (16) | 0.0767 (19) | 0.0122 (13)  | 0.0061 (14)  | -0.0112 (14) |
| C4  | 0.0784 (19) | 0.0382 (14) | 0.096 (2)   | 0.0072 (13)  | 0.0005 (17)  | -0.0028 (15) |
| C5  | 0.0590 (16) | 0.0446 (14) | 0.0783 (19) | -0.0029 (12) | 0.0079 (13)  | 0.0063 (13)  |
| C6  | 0.0394 (12) | 0.0392 (12) | 0.0550 (15) | 0.0021 (10)  | -0.0010 (10) | -0.0012 (11) |
| C7  | 0.0370 (12) | 0.0503 (14) | 0.0477 (14) | -0.0026 (10) | 0.0007 (9)   | 0.0011 (11)  |
| C8  | 0.0364 (12) | 0.0444 (13) | 0.0476 (14) | 0.0026 (10)  | -0.0039 (10) | -0.0026 (10) |
| C9  | 0.0472 (13) | 0.0401 (14) | 0.0506 (15) | 0.0028 (10)  | -0.0024 (10) | -0.0044 (11) |
| C10 | 0.0470 (14) | 0.0573 (16) | 0.0531 (16) | 0.0102 (12)  | -0.0031 (11) | -0.0058 (13) |
| C11 | 0.0676 (18) | 0.090 (2)   | 0.0663 (19) | 0.0167 (16)  | 0.0152 (14)  | -0.0173 (16) |
| C12 | 0.0520 (16) | 0.093 (2)   | 0.0642 (18) | 0.0203 (14)  | 0.0082 (13)  | -0.0162 (16) |
| C13 | 0.0517 (15) | 0.0826 (19) | 0.0598 (17) | 0.0113 (14)  | 0.0072 (12)  | -0.0058 (15) |
| C14 | 0.0605 (16) | 0.0681 (17) | 0.0635 (17) | 0.0112 (14)  | 0.0127 (13)  | -0.0002 (14) |
| C15 | 0.0484 (14) | 0.0513 (14) | 0.0690 (18) | 0.0042 (12)  | 0.0139 (12)  | -0.0074 (13) |
| C16 | 0.0617 (18) | 0.0521 (16) | 0.109 (2)   | 0.0106 (14)  | 0.0295 (16)  | 0.0079 (16)  |
| C17 | 0.0542 (18) | 0.0535 (17) | 0.143 (3)   | -0.0068 (14) | 0.0329 (18)  | -0.012 (2)   |
| C18 | 0.0490 (16) | 0.078 (2)   | 0.096 (2)   | 0.0006 (15)  | 0.0078 (15)  | -0.0310 (19) |
| C19 | 0.0669 (18) | 0.091 (2)   | 0.071 (2)   | 0.0004 (17)  | 0.0080 (15)  | 0.0048 (16)  |
| C20 | 0.0549 (16) | 0.0741 (19) | 0.080 (2)   | -0.0136 (14) | 0.0122 (14)  | 0.0070 (16)  |

### Geometric parameters ( $\text{\AA}$ , $^\circ$ )

|        |           |          |           |
|--------|-----------|----------|-----------|
| O1—C2  | 1.362 (3) | C11—H11A | 0.9600    |
| O1—C11 | 1.438 (3) | C11—H11B | 0.9600    |
| O2—C1  | 1.373 (3) | C11—H11C | 0.9600    |
| O2—C9  | 1.382 (3) | C12—C13  | 1.480 (4) |
| O3—C9  | 1.198 (3) | C12—H12A | 0.9700    |
| O4—C10 | 1.194 (3) | C12—H12B | 0.9700    |
| O5—C10 | 1.349 (3) | C13—C14  | 1.299 (4) |
| O5—C12 | 1.465 (3) | C13—H13A | 0.9300    |
| C1—C6  | 1.394 (3) | C14—C15  | 1.476 (4) |

|             |             |               |           |
|-------------|-------------|---------------|-----------|
| C1—C2       | 1.398 (3)   | C14—H14A      | 0.9300    |
| C2—C3       | 1.375 (4)   | C15—C16       | 1.379 (3) |
| C3—C4       | 1.389 (4)   | C15—C20       | 1.389 (4) |
| C3—H3A      | 0.9300      | C16—C17       | 1.383 (4) |
| C4—C5       | 1.370 (4)   | C16—H16A      | 0.9300    |
| C4—H4A      | 0.9300      | C17—C18       | 1.377 (4) |
| C5—C6       | 1.405 (3)   | C17—H17A      | 0.9300    |
| C5—H5A      | 0.9300      | C18—C19       | 1.370 (4) |
| C6—C7       | 1.428 (3)   | C18—H18A      | 0.9300    |
| C7—C8       | 1.351 (3)   | C19—C20       | 1.381 (4) |
| C7—H7A      | 0.9300      | C19—H19A      | 0.9300    |
| C8—C9       | 1.469 (3)   | C20—H20A      | 0.9300    |
| C8—C10      | 1.485 (3)   |               |           |
| C2—O1—C11   | 116.7 (2)   | H11A—C11—H11B | 109.5     |
| C1—O2—C9    | 122.93 (18) | O1—C11—H11C   | 109.5     |
| C10—O5—C12  | 116.4 (2)   | H11A—C11—H11C | 109.5     |
| O2—C1—C6    | 121.0 (2)   | H11B—C11—H11C | 109.5     |
| O2—C1—C2    | 117.3 (2)   | O5—C12—C13    | 111.3 (2) |
| C6—C1—C2    | 121.7 (2)   | O5—C12—H12A   | 109.4     |
| O1—C2—C3    | 126.0 (2)   | C13—C12—H12A  | 109.4     |
| O1—C2—C1    | 116.1 (2)   | O5—C12—H12B   | 109.4     |
| C3—C2—C1    | 117.9 (2)   | C13—C12—H12B  | 109.4     |
| C2—C3—C4    | 121.3 (3)   | H12A—C12—H12B | 108.0     |
| C2—C3—H3A   | 119.4       | C14—C13—C12   | 124.9 (3) |
| C4—C3—H3A   | 119.4       | C14—C13—H13A  | 117.6     |
| C5—C4—C3    | 120.9 (2)   | C12—C13—H13A  | 117.6     |
| C5—C4—H4A   | 119.6       | C13—C14—C15   | 127.8 (3) |
| C3—C4—H4A   | 119.6       | C13—C14—H14A  | 116.1     |
| C4—C5—C6    | 119.5 (2)   | C15—C14—H14A  | 116.1     |
| C4—C5—H5A   | 120.3       | C16—C15—C20   | 117.2 (3) |
| C6—C5—H5A   | 120.3       | C16—C15—C14   | 119.8 (3) |
| C1—C6—C5    | 118.8 (2)   | C20—C15—C14   | 123.0 (2) |
| C1—C6—C7    | 117.2 (2)   | C15—C16—C17   | 121.5 (3) |
| C5—C6—C7    | 124.0 (2)   | C15—C16—H16A  | 119.2     |
| C8—C7—C6    | 122.5 (2)   | C17—C16—H16A  | 119.2     |
| C8—C7—H7A   | 118.7       | C18—C17—C16   | 120.2 (3) |
| C6—C7—H7A   | 118.7       | C18—C17—H17A  | 119.9     |
| C7—C8—C9    | 119.6 (2)   | C16—C17—H17A  | 119.9     |
| C7—C8—C10   | 122.2 (2)   | C19—C18—C17   | 119.2 (3) |
| C9—C8—C10   | 118.1 (2)   | C19—C18—H18A  | 120.4     |
| O3—C9—O2    | 115.8 (2)   | C17—C18—H18A  | 120.4     |
| O3—C9—C8    | 127.6 (2)   | C18—C19—C20   | 120.3 (3) |
| O2—C9—C8    | 116.62 (19) | C18—C19—H19A  | 119.9     |
| O4—C10—O5   | 123.1 (2)   | C20—C19—H19A  | 119.9     |
| O4—C10—C8   | 125.8 (3)   | C19—C20—C15   | 121.5 (3) |
| O5—C10—C8   | 111.2 (2)   | C19—C20—H20A  | 119.3     |
| O1—C11—H11A | 109.5       | C15—C20—H20A  | 119.3     |
| O1—C11—H11B | 109.5       |               |           |

## supplementary materials

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|              |              |                 |             |
|--------------|--------------|-----------------|-------------|
| C9—O2—C1—C6  | 4.0 (3)      | C7—C8—C9—O3     | -179.1 (2)  |
| C9—O2—C1—C2  | -174.71 (19) | C10—C8—C9—O3    | -1.1 (3)    |
| C11—O1—C2—C3 | 2.6 (4)      | C7—C8—C9—O2     | 1.1 (3)     |
| C11—O1—C2—C1 | -177.7 (2)   | C10—C8—C9—O2    | 179.11 (18) |
| O2—C1—C2—O1  | -0.4 (3)     | C12—O5—C10—O4   | -1.3 (3)    |
| C6—C1—C2—O1  | -179.15 (19) | C12—O5—C10—C8   | 178.20 (18) |
| O2—C1—C2—C3  | 179.3 (2)    | C7—C8—C10—O4    | 172.9 (3)   |
| C6—C1—C2—C3  | 0.6 (3)      | C9—C8—C10—O4    | -5.0 (4)    |
| O1—C2—C3—C4  | 179.8 (2)    | C7—C8—C10—O5    | -6.5 (3)    |
| C1—C2—C3—C4  | 0.0 (4)      | C9—C8—C10—O5    | 175.55 (18) |
| C2—C3—C4—C5  | -0.1 (4)     | C10—O5—C12—C13  | -83.6 (3)   |
| C3—C4—C5—C6  | -0.5 (4)     | O5—C12—C13—C14  | -115.1 (3)  |
| O2—C1—C6—C5  | -179.8 (2)   | C12—C13—C14—C15 | -179.5 (3)  |
| C2—C1—C6—C5  | -1.2 (3)     | C13—C14—C15—C16 | -164.1 (3)  |
| O2—C1—C6—C7  | -1.3 (3)     | C13—C14—C15—C20 | 15.9 (4)    |
| C2—C1—C6—C7  | 177.37 (19)  | C20—C15—C16—C17 | -3.4 (4)    |
| C4—C5—C6—C1  | 1.1 (4)      | C14—C15—C16—C17 | 176.6 (3)   |
| C4—C5—C6—C7  | -177.4 (2)   | C15—C16—C17—C18 | 2.2 (4)     |
| C1—C6—C7—C8  | -1.3 (3)     | C16—C17—C18—C19 | 0.7 (4)     |
| C5—C6—C7—C8  | 177.1 (2)    | C17—C18—C19—C20 | -2.3 (4)    |
| C6—C7—C8—C9  | 1.4 (3)      | C18—C19—C20—C15 | 0.9 (4)     |
| C6—C7—C8—C10 | -176.50 (19) | C16—C15—C20—C19 | 1.9 (4)     |
| C1—O2—C9—O3  | 176.3 (2)    | C14—C15—C20—C19 | -178.2 (3)  |
| C1—O2—C9—C8  | -3.8 (3)     |                 |             |

### Hydrogen-bond geometry ( $\text{\AA}$ , $^\circ$ )

| $D-H\cdots A$                      | $D-H$ | $H\cdots A$ | $D\cdots A$ | $D-H\cdots A$ |
|------------------------------------|-------|-------------|-------------|---------------|
| C4—H4A $\cdots$ O3 <sup>i</sup>    | 0.93  | 2.51        | 3.429 (3)   | 170           |
| C17—H17A $\cdots$ O4 <sup>ii</sup> | 0.93  | 2.44        | 3.294 (4)   | 153           |

Symmetry codes: (i)  $x, y+1, z$ ; (ii)  $-x+1, y+1/2, -z+1/2$ .



Fig. 1

